Admission test -Degree Course in Pharmacy, University of Rome 'Tor Vergata', Academic Year 2017/2018

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When the anion MnO<sup>4</sup> is transformed into cation Mn<sup>2+</sup>, manganese:
1
Α
        Goes from oxidation number +7 to +2 and is reduced
        Goes from oxidation number +7 to +2 and is oxidized
В
C
        Goes from oxidation number -1 to +2 and is reduced
        Goes from oxidation number -2 to +2 and is reduced
D
E
        None of the above
2
        Indicate which is the sulfite ion:
A
        SO_4^{2-}
В
        SO_3^{2-}
C
        S^{2+}
D
        S_2O_3^{2-}
E
3
        Given that the atomic mass of carbon is 12, how many atoms are present in 0,36g of carbon?
A
В
        0.03
        6.022 x 1023
\mathbf{C}
        1,8066 x 1022
D
E
        12
        If in a water solution the concentration of H<sub>3</sub>O<sup>+</sup> ions is 0,001M, what is the pH?
4
A
        1
        0,001
В
C
        3
        30
D
E
5
        How many grams of ethyl alcohol (C<sub>2</sub>H<sub>5</sub>OH, MW 46,1) are present in 500mL of a solution 0,2M?
A
        4,61g
        6,022 x 1023g
В
C
        1,0 x 1023g
D
        0,461g
        None of the above
E
6
        Methanol is:
Α
        An organic compound containing a hydroxy group
В
        An aromatic organic compound
C
        A halogenated organic compound
D
        An organic compound containing an amino group
E
        None of the above
7
        An atom that contains 19 protons, 20 neutrons and 19 electrons has the following mass number:
A
В
        20
C
        39
D
        58
E
        19,5
8
        Which of the following solutions has pH>7?
A
        50mL of a 0,1M solution of NaCl
        50mL of a 0,1M solution of HCl
В
C
        250mL of a 0.1M solution of HCl
D
        50mL of a 0,1M solution of NaOH
E
        50mL of a 0,1M solution of NH<sub>4</sub>Cl
9
        A 0,5 M solution of KCl contains:
A
        1mol of solute per liter of solution
        0,5g of solute per liter of solution
В
        0,5g of solute per Kg of solution
C
D
        1mole of solute per Kg of solution
        0,5 moles of solute per liter of solution
E
10
        Which of the following couples of atoms can form a covalent bonding?
        H: Na
Α
        H; Br
В
C
        Na: Br
D
        Ca; Cl
Ε
        Ca; O
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11 A gas sample at $T=0^{\circ}C$ and P=1 atm has volume V=44.8 L, contains: About 1 mole

- A About I IIIOIE
- B About 1gram
- C About 2 moles
- D About 2grams
- E About 2 molecules
- 12 Ice and water constitute a system:
- A Chemically heterogeneous and physically heterogeneous
- B Chemically homogeneous and physically heterogeneous
- C Chemically homogeneous and physically homogeneous
- D Chemically heterogeneous and physically homogeneous
- E None of the above

Which species can be formed upon combustion of methane (CH₄)?

- A Hydrogen and oxygen
- B Carbon and hydrogen
- C Propane
- D Methane gas
- E Carbon dioxide and water

14 In order to neutralize 100mL of a 0,01M solution of KOH it is necessary to add:

- A 0,1mL of H₂O
- B 0,1mL of a 1 M solution of NaOH
- C $1 \text{mL of a } 10^{-2} \text{ M solution of HNO}_3$
- D 100mL of a 10⁻² M solution of HNO₃
- E 100mL of a 10⁻² M solution of NaOH

15 In order to neutralize 100ml of a10⁻³M solution of HCl it is necessary to add:

- A 1000ml of H₂O
- B 10ml of a 0.01 M solution of NaOH
- C 10ml of a 0.01 M solution of KCl
- D 1ml of a 10⁻² M solution of HNO₃
- E 1ml of a 10⁻² M solution of NaOH

16 In eukaryotic cells, DNA occurs only in:

- A Cell nuclei, mitochondria, Golgi apparatus
- B Mitochondria, Golgi apparatus, chloroplasts
- C Cell nuclei, lysosomes, chloroplasts
- D Cell nuclei, mitochondria, chloroplasts
- E Mitochondria, chloroplasts, lysosomes

17 Two parents who do not exhibit phenylketonuria (PKU) have a son with PKU. Which of the following conclusions can be drawn from this situation?

- A The allele for PKU is located on the Y chromosome
- B PKU is a dominant trait
- C PKU is a recessive trait
- D A mutation occurred in the sperm of the father
- E None of the above

Which of the following cellular processes normally produces ATP from glucose in the absence of oxygen?

- A Krebs cycle
- B Glycolysis
- C Chemiosmosis
- D Calvin cycle
- E None of the above

Which of the following best describes the pathway of a protein from its manufacture to its secretion from the cell?

- A Endoplasmic reticulum → Golgi complex → secretory vesicle
- B Secretory vesicle → endoplasmic reticulum → Golgi complex
- C Secretory vesicle → Golgi complex → endoplasmic reticulum
- D Golgi complex → endoplasmic reticulum → secretory vesicle
- E None of the above

A person touches a hot object and immediately moves her finger away from it. Which of the following structures is the first to receive an impulse triggered by the stimulus?

- A Synapse
- B Ventral root ganglion
- C Motor neuron
- D Sensory neuron
- E None of the above

21 The sympathetic nervous system Α Increases heart rate В Is voluntary C Is activated mainly in sleep D Activates digestion Ε None of the above 22 A health disorder that results from a deficiency of thyroid hormone production in adults and is characterized by a low metabolic rate, increase in body weight and tiredness is: A Hyperthyroidism В Goiter C Cretinism Hypothyroidism D Ε None of the above 23 In eukaryotic cells, protein synthesis takes place at the: A Nucleus В Mitochondria C Chloroplasts Ribosomes D Ε Lysosome 24 The four bases which form the code words for DNA are: UTAC A **ACTU** В C **AGCU** D **ATCG** Ε None of the above 25 Prokaryotic cells are different from eukaryotic cells because prokaryotic cells: A Are much bigger В Have no cell nuclei C Have cell nuclei D Have no cell wall E Have mitochondria and chloroplasts 26 The DNA double helix is stabilized by hydrogen bonds between: Purine bases and complementary pyrimidine bases Α В Pyrimidine bases only Purine bases only C D Molecules of deoxyribose and phosphate groups Ε Nitrogenous bases and phosphate groups 27 During which of the following stages of cell cycle DNA replication occurs? Early prophase A G1 phase В G2 phase C G0 phase D Ε S phase 28 Which of the following statements best describes mitosis? A It occurs only in the reproductive structure of organism During this stage DNA is synthesized В It is one of the interphase stages C It is the division of mother cell into two daughter cells, genetically identical to each other and to their parent cell D Ε It is the division of mother cell into two daughter cells, genetically different from each other and from their parent cell 29 Gene expression may be regulated: Α Only during translation В Only during transcription C During transcription and translation Only during replication D Ε During replication and transcription 30 One of a series of different versions of gene is called: Genome

A B

C

D

Ε

Haploid

Diploid

Allosome

Allele

31 Which of the following processes can decrease the genetic variation in natural populations? A Recombination В Immigration C Mutagenesis D Inbreeding Ε Hybridization 32 **Eutrophication is the process of:** Vegetation succession in lakes A Nutrient depletion of lakes В C Increase inflow of minerals in lakes D Primary succession in lakes Ε Biological renewal of aquatic ecosystems Which of the following types of plants tissue transport water and minerals? 33 A Phloem В Xylem C Meristem D Vascular cambium **Epidermis** Ε 34 Which of the following require a host cell because they are notable to make proteins on their own? Paramecium caudatum A Escherichia coli В Influenza virus C D Saccharomyces cerevisiae E Euglena viridis 35 Which of the following is found in an unusual high proportion in cells of cardiac muscle? A Lysosomes В Mitochondria C **MRNA** D Golgi apparatus E Cell nucleus **36** Which of the following organisms conduct ethanol fermentation? A Yeasts В Bears C Spiders Barley D Ε Mushrooms **37** Which of the following is the right order of the four stages of food processing? Digestion, ingestion, absorption, elimination A Ingestion, digestion, elimination, absorption В C Ingestion, digestion, absorption, elimination Digestion, ingestion, absorption, elimination D Absorption, elimination, digestion, ingestion Ε 38 Which hormone is produced by the placenta? A Progesterone В Insulin Thyroxin C D Melatonin Ε None of the above 39 Which of the following is not characteristic for mammals? A Mammary glands В Hair C Cold-blooded D Three middle ears bones Ε Sweat glands 40 All of the members of a particular species that live in one area are called a(an): A Biotope Community В Ecosystem \mathbf{C} D Biosphere Population Ε

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41
         Consider the following inequalities:
A
         (2/3)>(3/4)
В
         (2/3)<(2/5)
        z^{-1} > z^{-2} when z is real and z>1
C
         2 + \frac{1}{10} > 1 + \frac{11}{10}
D
Е
         None of the above
42
         What is the solution of the linear system x+y=5; 2x+3y=4?
A
         x=1, y=1
         x=11, y=-6
В
C
         The linear system has infinite solutions
D
         x=-11, y=6
Ε
         The linear system has no solution
         What is/are the correct value(s) of (a + b)^2(a, b \ real \ numbers)?
43
Α
         a^2 + ab + b^2
         a^2 + b^2
В
        a^2 - b^2
a^2 + 2ab + b^2
C
D
         a^2 + 2ab + b^2
Ε
        Consider the equation 4 - \sqrt{7 + \sqrt{9 + \sqrt{4 + x}}} = 0. Mark the correct value of x
44
         3660
A
В
         46
C
         567
D
         68
Ε
         5180
         What is/are the correct value(s) of x for the equation log_x(16) = \frac{4}{3}?
45
         x = \sqrt[3]{65536}
A
В
         x = 64
C
         x = 2
D
         x = 8
E
         x = 1
         Calculates the mass of the motorcycle with speeds of 10 m/s and kinetic energy of 10000 J
46
Α
         100 \text{ kg}
В
         300 kg
C
         200 kg
D
         400 kg
Ε
         500 kg
47
         Which physical quantity can be measured in Newton meter (N m)?
A
В
         Momentum
        Kinetic energy
C
D
         Power
Ε
         Work
48
         Which physical quantity can be measured in Newton second (N s)?
A
         Force
В
         Momentum
C
         Kinetic energy
D
         Power
Ε
         Work
49
         When do you have a uniform circular motion?
Α
         When there is a variable centripetal force
В
         When there is a constant centripetal force
C
         When there is a constant centrifugal force
D
         When there is a variable centrifugal force
Ε
         When there is a constant velocity
50
         A net force FA acts on object A, and a net force FB acts on object B. The mass of object B is twice the mass of object A,
         and the acceleration of object B is twice that of object A. Which of the following relations between FA and FB is true?
A
         FB = \frac{1}{4} FA
В
         FB = 1/2FA
C
         FB = FA
         FB = 2FA
D
         FB = 4 FA
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E

51 In which year the Wright brothers achieved the first powered, sustained and controlled airplane flight? Α 1918 В C 1880 D 1903 Ε None of the above 52 The Machu Picchu is an archeological site located in? Α Perù В Iran C Mexico D Colombia Ε None of the above 53 Which is the Capital of Romania? A Budapest Bucarest В \mathbf{C} Belgrad D Novi Sad E None of the above 54 Which architect designed the Sagrada Familia in Barcellona? Α Le Corbusier В Carlos Gardel Antoni' Gaudi' C Oscar Niemeyer D Ε None of the above Who first discovered the oral polio vaccine? 55 A Fleming В Sabin C Dulbecco D Pasteur E None of the above **56** In which year Christopher Columbus discovered America? A 1592 В 1472 C 1492 Ε 1542 57 Who designed the cupola of Santa Maria del Fiore in Florence? Michelangelo A Raffaello В C Maderno D Brunelleschi E None of the above **58** Which of the following movies was not directed by Stanley Kubrick? Α 2001: A Space Odyssey В Lawrence of Arabia \mathbf{C} Shining D A clockwork Orange E None of the above **59** Which of the following Nations is not a Republic? Α Portugal Austria В C Greece D Sweden E None of the above 60 Roger Federer is: Football player Α Basketball player В

 \mathbf{C}

D

E

Volleyball player

Tennis player None of the above